## Room temperature complete reduction of nitroarenes over novel Cu/SiO<sub>2</sub>@NiFe<sub>2</sub>O<sub>4</sub> nano-catalyst in aqueous medium – A kinetic and mechanistic study

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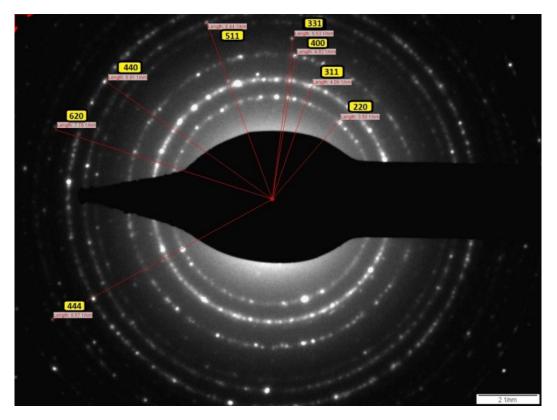


Fig. S1. ED indexing of NiFe<sub>2</sub>O<sub>4</sub>.

### EDS data

## MP1.1

Spectrum processing :

Peak possibly omitted : 0.271 keV

Processing option : All elements analyzed (Normalised)

Number of iterations = 3 Standard :

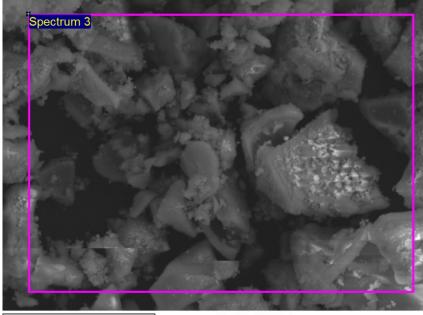
O SiO2 1-Jun-1999 12:00 AM

Si SiO2 1-Jun-1999 12:00 AM

Fe Fe 1-Jun-1999 12:00 AM

Ni Ni 1-Jun-1999 12:00 AM

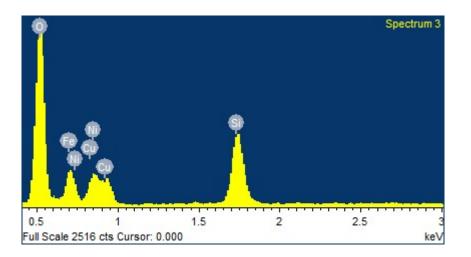
Cu Cu 1-Jun-1999 12:00 AM



20µm

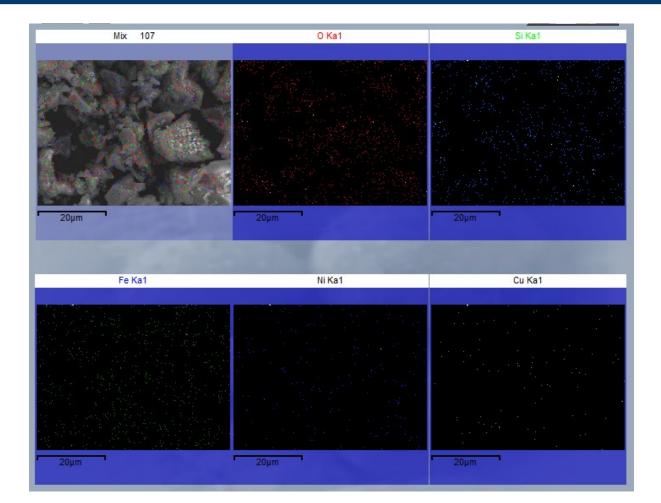
Electron Image 1

Element	Weight%	Atomic%
ОК	32.91	61.19
Si K	8.20	8.68
Fe K	29.85	15.90
Ni K	16.35	8.28
Cu L	12.69	5.94
Totals	100.00	
	l	



## **Elemental Mapping**

The elemental mapping data showing uniform distribution of Si and Cu on the support.



## MP1.1

### **XPS of individual elements**

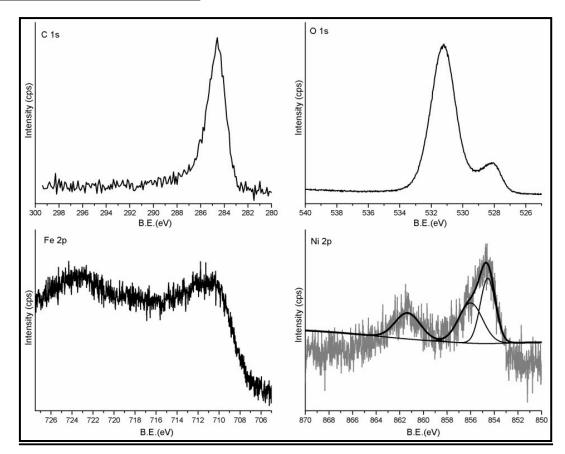


Fig. S2. XPS spectrum showing C 1s peak calibrated to 284.6eV. Remaining spectra are of individual elements present in the active catalyst sample Cu/SiO<sub>2</sub>@NiFe<sub>2</sub>O<sub>4</sub>. Starting from left C 1s, O 1s, Fe 2p, Ni2p.

## <u>VSM</u>

#### Table S1

Magnetisation data of the samples

Sample	Ms(emu/g)	Mr (emu/g)	Hc (Oe)
Cu/SiO <sub>2</sub> @NiFe <sub>2</sub> O <sub>4</sub>	22.1165	2.6510	232
SiO <sub>2</sub> @NiFe <sub>2</sub> O <sub>4</sub>	24.4882	2.3524	200
NiFe <sub>2</sub> O <sub>4</sub>	28.7444	2.5489	132

Ms- Saturation Magnetization

Mr- Remanent Magnetization

Hc - Coercivity

As expected the saturation magnetisation (Ms) was highest for NiFe<sub>2</sub>O<sub>4</sub> which decreased after silica coating and was found lowest for Cu/SiO<sub>2</sub>@NiFe<sub>2</sub>O<sub>4</sub> as seen in the fig. S6.

#### **Product charaterisation**

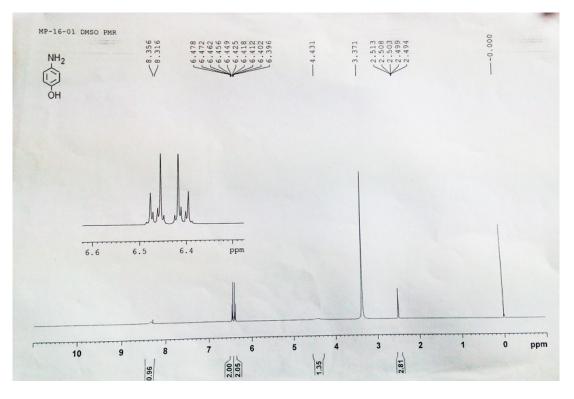
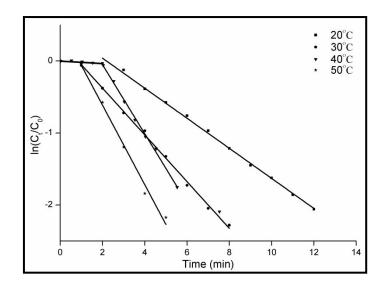


Fig. S3. <sup>1</sup>H NMR spectrum of the isolated product when performed on 1 mmol scale matches with reference <sup>1</sup>.

### Reference

1 T. Aditya and T. Pal, *Chem. Commun.*, 2015, **51**, 9410–9431.



**Fig. S4.** Plot of *k* at different temperatures.

#### Table S2

Recyclability of the catalyst

Cycle	Time (min) for total		
	conversion		
1 <sup>st</sup>	7		
2 <sup>nd</sup>	7		
3 <sup>rd</sup>	11		
4 <sup>th</sup>	14		
5 <sup>th</sup>	20		

#### Table S3

Solvent Studies using Nitrobenzene (0.5 mmol), Catalyst (6 mg), NaBH4 (2.5 mmol)

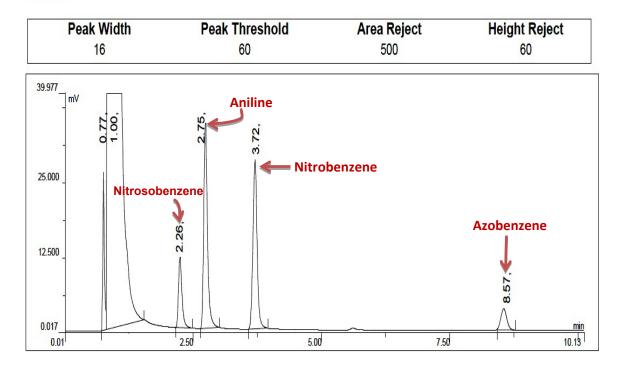
Entry	Solvent system	Observations*
1	THF	no desired product
2	Acetonitrile	no desired product
3	Methanol	3.5h for complete conversion
4	Ethanol	5h for complete conversion
5	Water	2.5h for complete conversion
6	1:1 Water-THF	8h for complete conversion
7	1:1 Water-Acetonitrie	Multiple spots even after 5h
8	1:1 Water-Ethanol	Over 7h for complete conversion
9	1:1 Water-Methanol	2h for complete conversion

\* Reactions were monitored on TLC.

### GC CHROMATOGRAMS

 Data File :
 E:\Mira\nitro\rxn\nitrobenzene\nitrobenz+anilin+azobenz+nitroso std 120-160-180-220-260.Dat

Method File :			
Sample Name :	Sample1	Analysis Type :	Percent Method
Detector :	FID	Time :	9:18:17 PM
System :	GC		0.02259311
Run Date :	8/5/2016	Chan No :	Chan 2



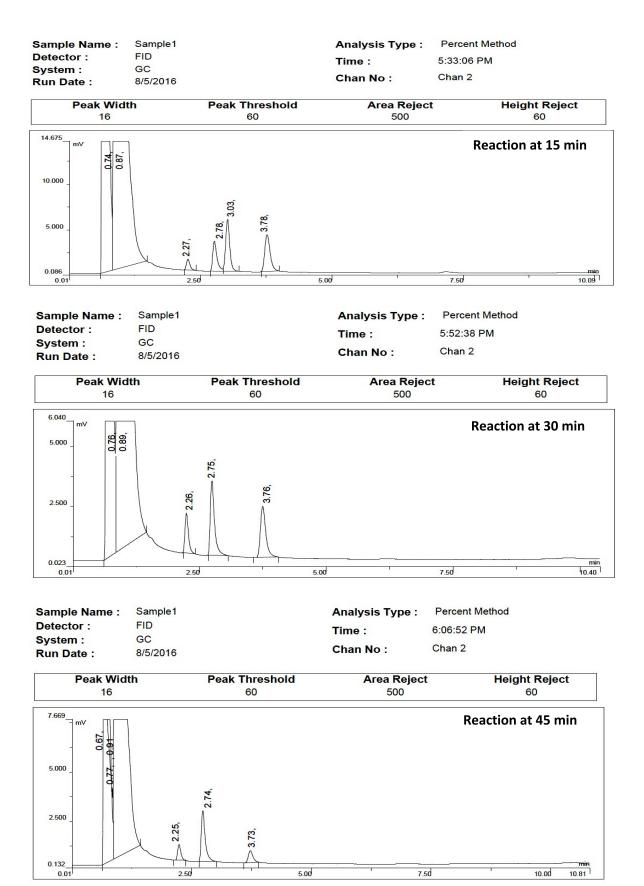
**Fig. S5.** A representative GC chromatogram showing Rt values of reactant Nitrobenzene, product aniline, and two intermediates Nitrosobenzene and Azobenzene.

All the GC runs were carried out using Ovi 101 column and same sequencer settings.

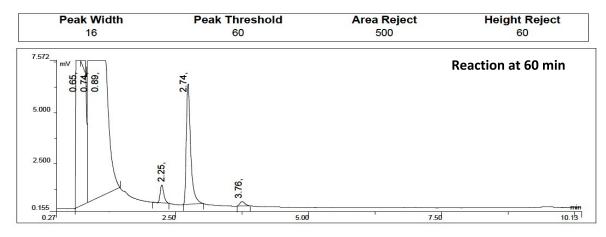
The above chromatograph displays the probable components of Nitrobenzene reduction. Authentic samples were injected individually to determine the retention time of the components with the set program. The subsequent chromatographs show the progress of the reaction with time in presence of  $Cu/SiO_2@NiFe_2O_4$ .

\*<u>NOTE</u>: Since the boiling point of aniline (179°C) and the internal standard used n-decane (174 °C) are very closed and it couldn't be well resolved at the set program, internal standard wasn't used while performing the mechanistic studies.

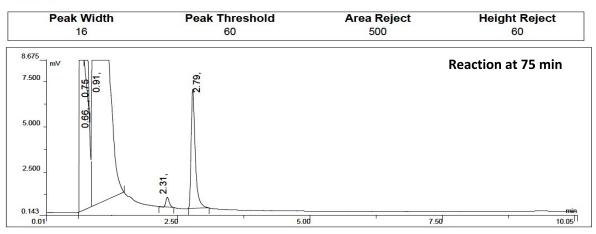
#### Mechanistic Study using Nitrobenzene as substrate over time



Sample Name :	Sample1	Analysis Type :	Percent Method
Detector :	FID	Time :	6:21:29 PM
System :	GC		
Run Date :	8/5/2016	Chan No :	Chan 2



Sample Name :	Sample1	Analysis Type :	Percent Method
Detector :	FID	Time :	6:34:37 PM
System : Run Date :	GC 8/5/2016	Chan No :	Chan 2



Sample Name :	Sample1	Analysis Type :	Percent Method
Detector :	FID	Time :	6:49:15 PM
System :	GC		Oh an O
Run Date :	8/5/2016	Chan No :	Chan 2

